



CAVITATION DAMAGE TO POLYPROPYLENE PIPING IN HIGH- RISE DHWR APPLICATIONS

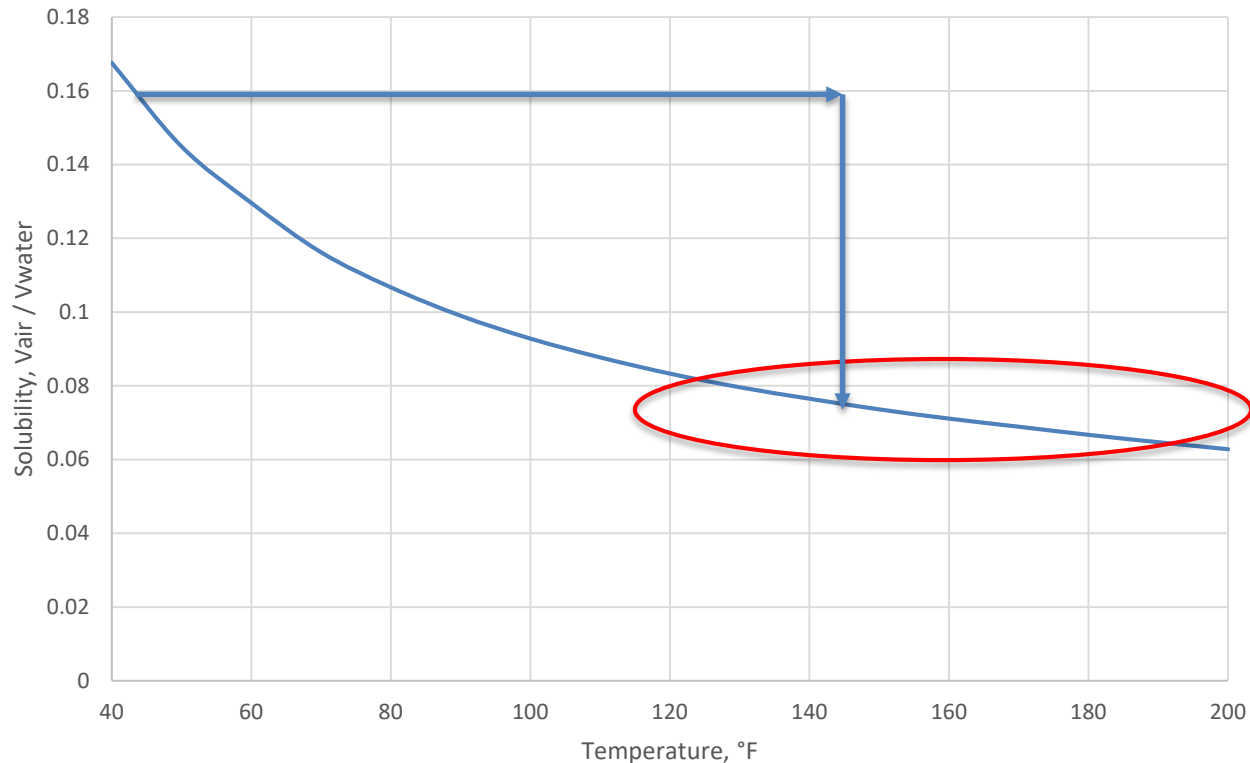
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Topics

- Temperature
 - Air/oxygen, Total Dissolved Gas (TDG) content
 - Boiler conditions
 - Saturation, source water
- Pressure
 - Air/oxygen, Total Dissolved Gas (TDG) content
 - Strength vs. temperature relationship
 - Water hammer, pressure surge and low pressure, transient
 - Cavitation
- Flow
 - Hydraulic balancing
 - Flow restriction
 - Velocity changes, water hammer
- Water Chemistry
 - Oxidation, pH, ORP
 - Sources of oxidizing chemicals
 - Accelerating effects of temperature, catalysts
- Cavitation damage examples
 - Domestic hot water recirculating systems
 - Mechanical damage, crack initiation
 - Stabilizer depletion, crack propagation

Temperature

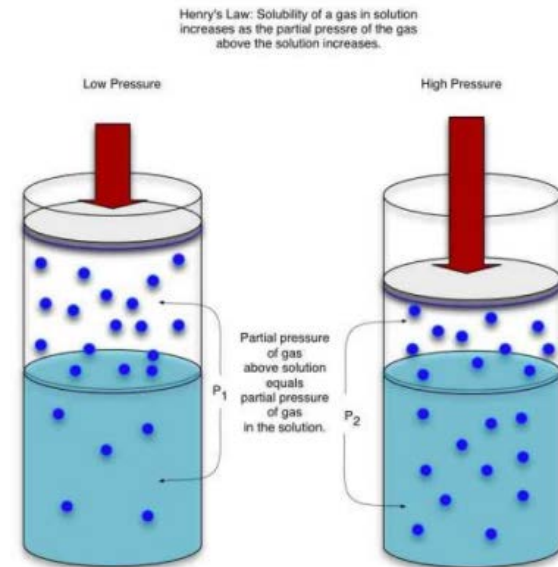
Air solubility in water at 80 psig



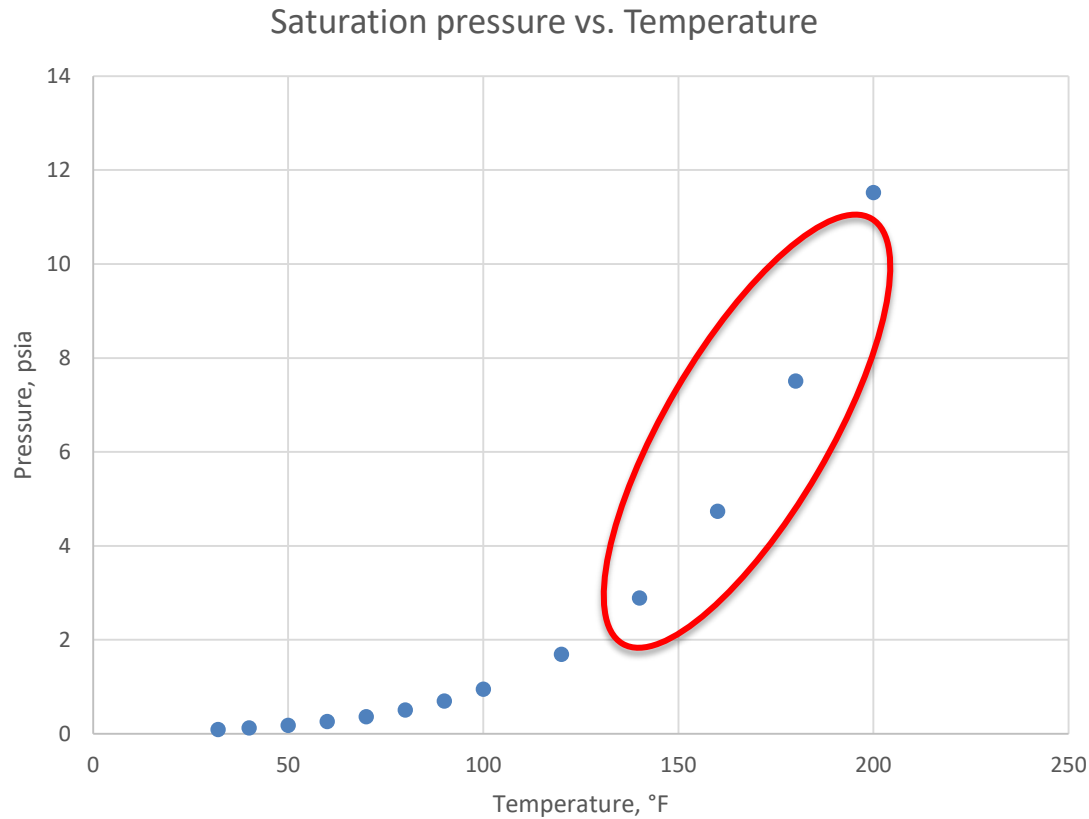
- As temperature increases, dissolved air comes out of solution
- For plumbing systems, incoming cold water is normally close to saturation

Temperature

- Gaseous cavitation
- Pressure falls below saturation pressure of fluid
- As temperature increases, saturation pressure decreases
- Release of gas, less dissolved gas in fluid, higher gas concentration out of solution



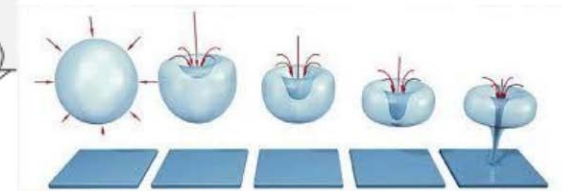
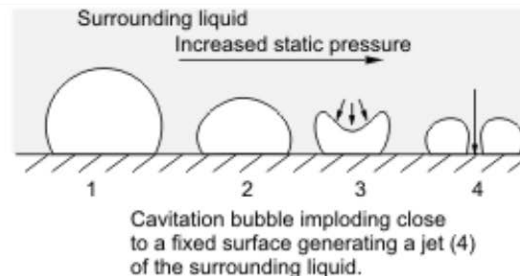
Temperature



- As temperature increases, vapor pressure increases
- Increases likelihood of vapor bubbles forming in low pressure region

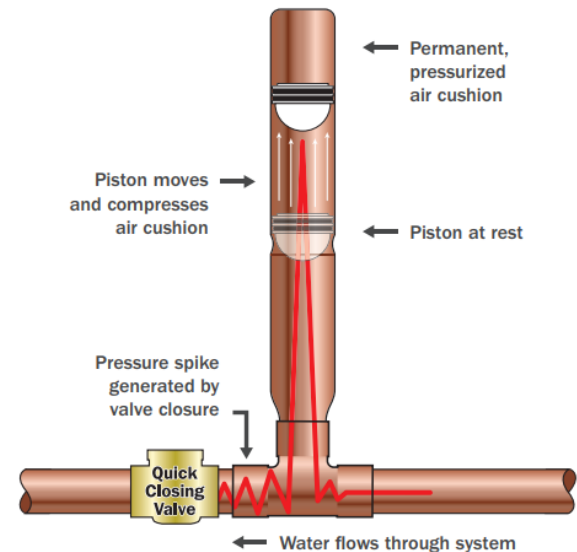
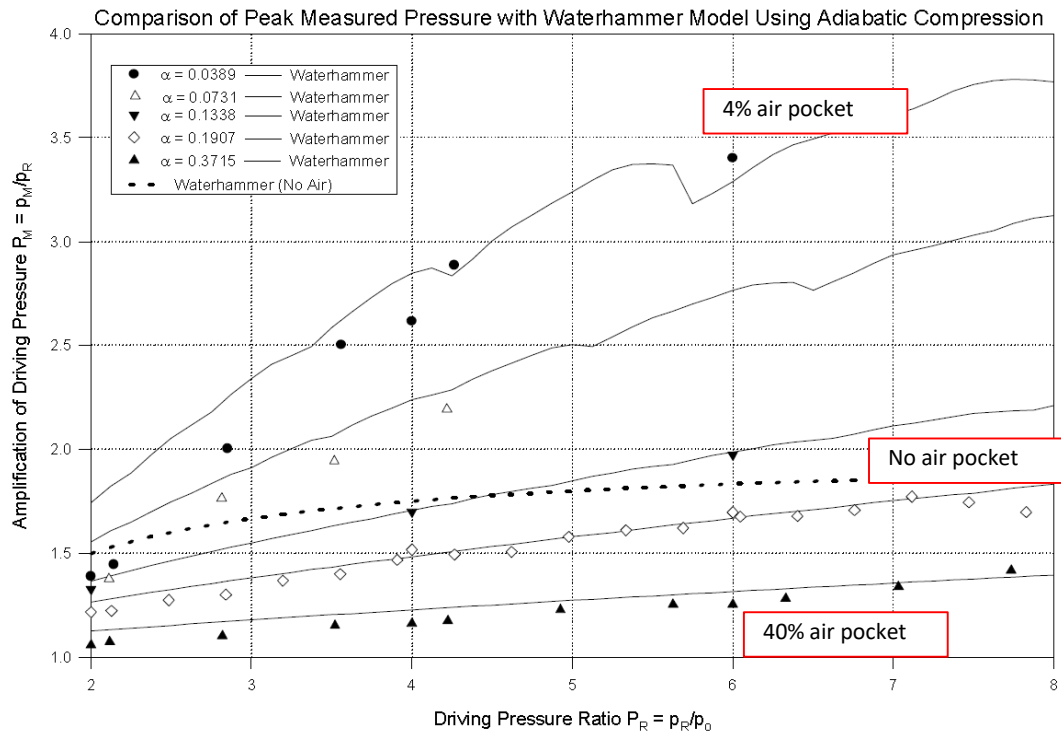
Temperature

- Vaporous cavitation
- Pressure falls below vapor pressure of fluid
- As temperature increases, vapor pressure increases
- Locally low static pressure will promote the production of vapor bubble



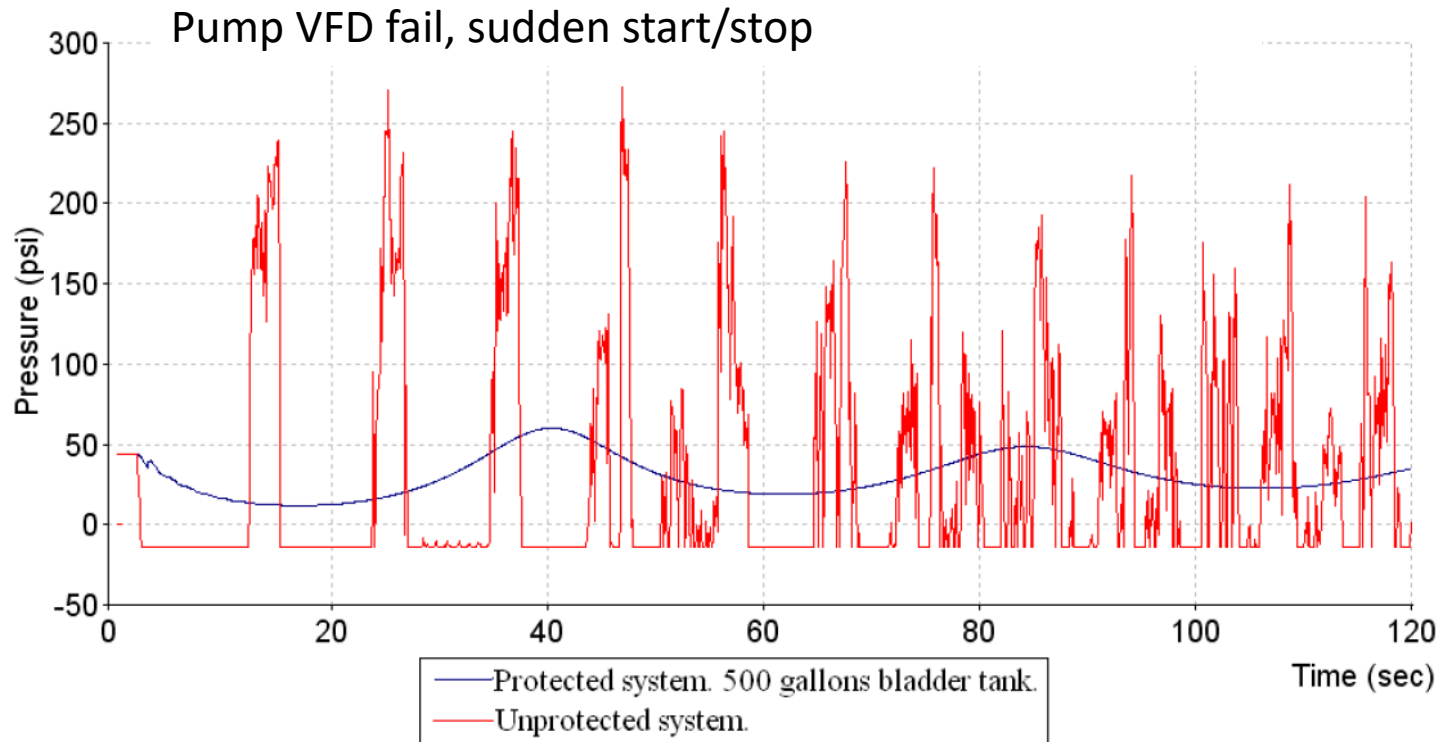
Pressure

- Pressure surge, water hammer
- If air in pipelines can substantially magnify the pressure surge effects, why do water hammer arresters work?



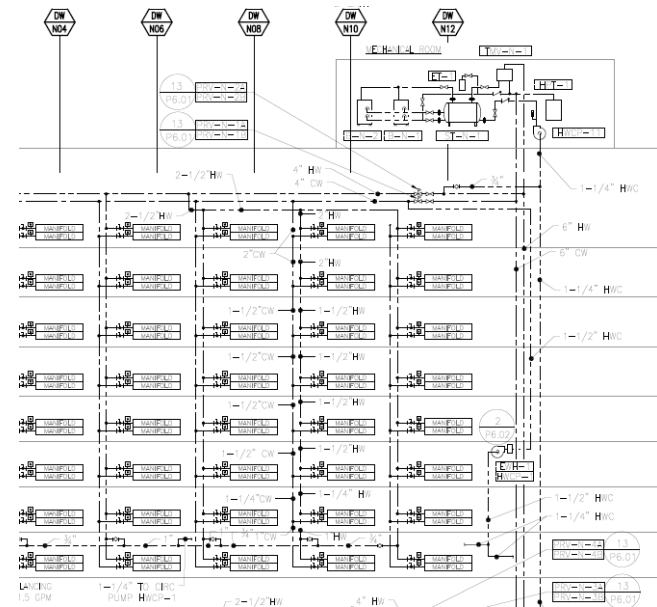
Pressure

- Conversely, transient pressure surges, water hammer will also cause low pressure events
- Cavitation (low pressure)



Flow

- Balanced flow is critical for:
 - Maintaining proper temperatures,
 - System operation at best efficiency point (energy savings), and
 - Reducing or eliminating damage to piping.
- Common in hydronic heating/cooling systems, information and data readily available
- Much less common in plumbing hot water circulating systems
- Balancing valves
 - Flow restrictors (flow balancing)
 - Temperature maintenance (thermal balancing, flow modulation)
- Oversized circulating pumps, or flow imbalance
 - High velocity in some legs, low or no flow in others
 - High velocity areas can erode/corrode piping, introduce copper into water



- Copper erosion/corrosion, pinhole leaks
 - Often initiated by cavitation, then susceptible to erosion/corrosion

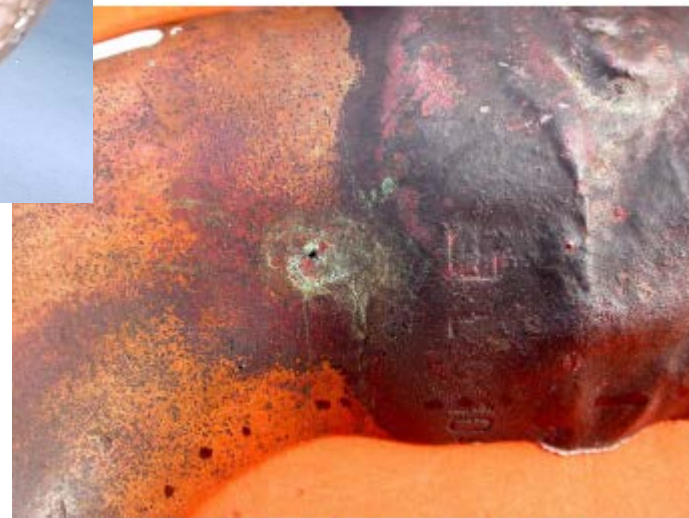


Figure 3. Pinhole leaks at head, viewed from outside of pipe.

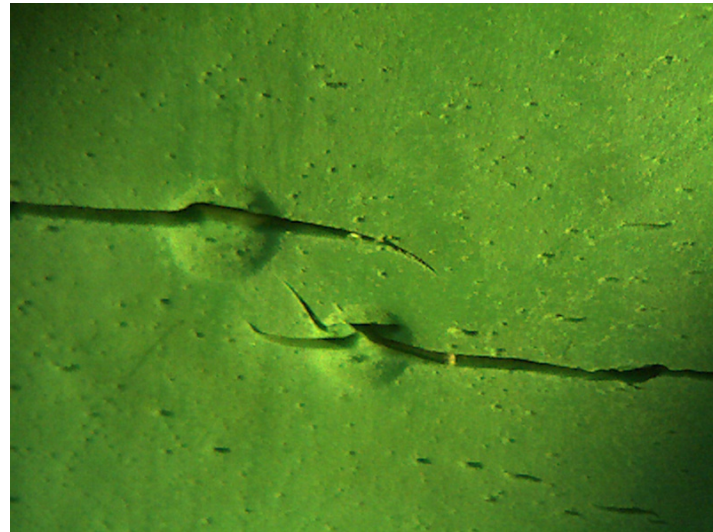
Identification

- Mechanical stress under elevated temperature, over-tightened clamp
- Cracks, degradation at clamp location
- Thermal oxidation, excessive temperature,

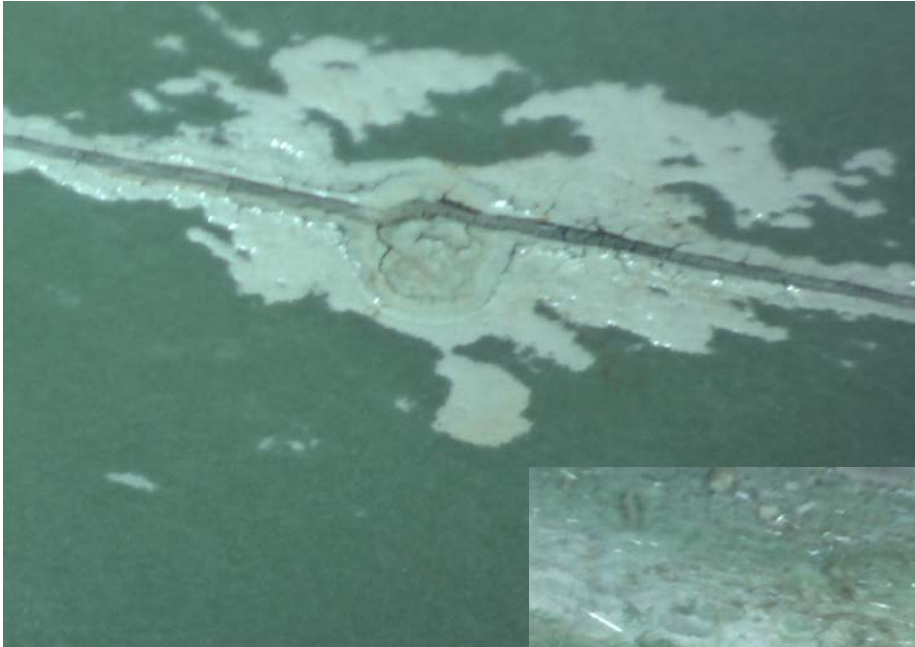


Identification

- Cavitation-initiated cracks, degradation
- Cavitation “divot” at crack origin
- Localized excessive stress, stabilizer depletion susceptible to crack propagation and oxidative degradation



Identification



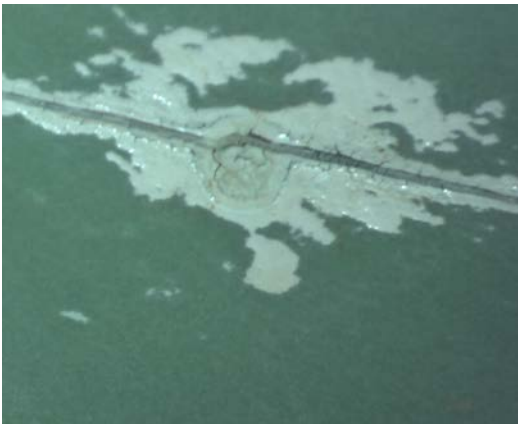
Divot with localized oxidative degradation

Cross-section, crack arrest lines



Identification

Sample	Wall depth	OIT, minutes
At divot location	ID (1 mm)	0.3
	Mid-wall	21.6
	OD (1 mm)	34.8
Undamaged area	ID (1 mm)	16.0
	Mid-wall	19.1
	OD (1 mm)	40.7



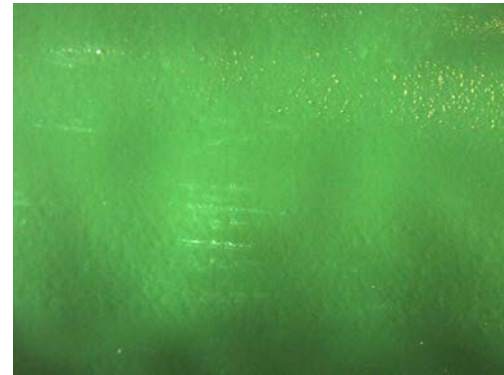
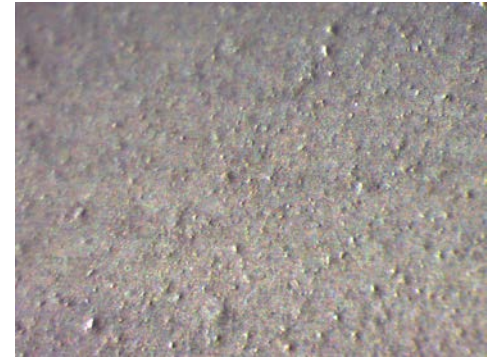
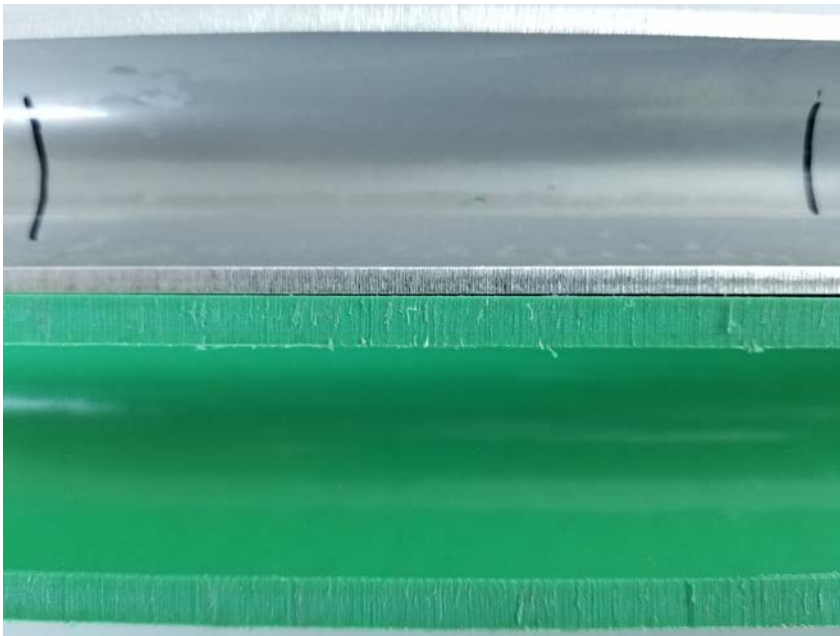
Testing

- Standard cavitation test stand, valving to reduce pressure on suction side of pump
- Aluminum pipe for comparison



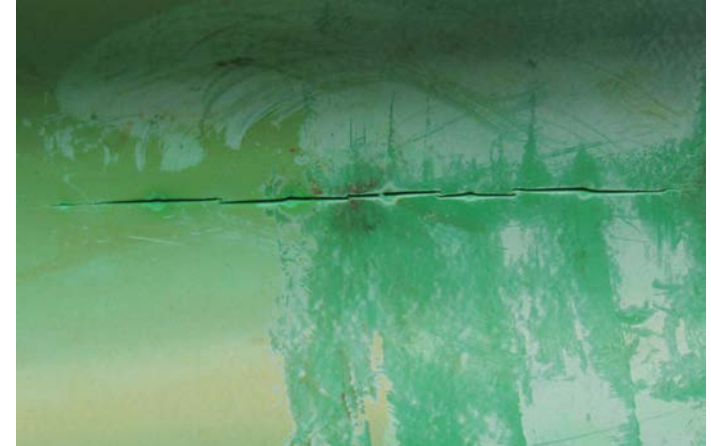
Testing

- Aluminum pipe cavitation damage
- No apparent damage to PP-R pipe



Testing

- Next Steps
 - No apparent damage to PP-R pipe
 - Damaged pipe in field
 - Differences in flow restriction vs. surge for cause of low pressure
 - Heat aging, oxidation, evaluation of surface embrittlement
- Operation and Maintenance
 - Air release
 - Pump operation, water hammer
 - Temperature monitoring
 - Copper corrosion, levels in water



QUESTIONS?